

(from Cornell Chem Dept; sing to "99 bottles")

Potassium, sodium, and ammonium salts,  
whatever they may be,  
can always be depended on  
for solubility.

when asked about the nitrates  
the answer is always clear,  
they each and all are soluble,  
is all we want to hear

most every chloride's soluble  
at least we've always read  
save silver, mercurous mercury  
and (slightly) chloride of lead

every single sulfate  
is soluble, 'tis said  
'cept barium and strontium  
and calcium and lead

hydroxides of metals won't dissolve  
that is, all but three  
potassium, sodium and ammonium  
dissolve quite readily

and then you must remember  
that you must not "forget"  
calcium, barium, strontium  
dissolve a little bit

the carbonates are insoluble, it's lucky that it's so,  
or else, our marble buildings  
would melt away like snow.

(repeat with feeling)  
Potassium, sodium, and ammonium salts,  
whatever they may be,  
can always be depended on  
for solubility.