

... Make That Eighteen Years

110. Solutions of tin (II) chloride and iron (III) chloride are mixed.
111. Solutions of cobalt (II) nitrate and sodium hydroxide are mixed.
112. Ethene gas is burned in air.
113. Equal volumes of equimolar solutions of phosphoric acid and potassium hydroxide are mixed.
114. Solid calcium sulfite is heated in a vacuum.
115. Excess hydrochloric acid is added to a solution of diamminesilver (I) nitrate.
116. Solid sodium oxide is added to distilled water.
117. A strip of zinc is added to a solution of 6.0-molar hydrobromic acid.
118. Calcium oxide powder is added to distilled water.
119. Solid ammonium nitrate is heated to temperatures above 300°C.
120. Liquid bromine is shaken with a 0.5 M sodium iodide solution.
121. Solid lead(II) carbonate is added to a 0.5 M sulfuric acid solution.
122. A mixture of powdered iron(III) oxide and powdered aluminum metal is heated strongly.
123. Methylamine gas is bubbled into distilled water.
124. Carbon dioxide gas is passed over hot solid, sodium oxide.
125. A 0.2 M barium nitrate solution is added to an alkaline 0.2 M potassium chromate solution.
126. A small piece of calcium metal is added to hot distilled water.
127. Butanol is burned in air.
128. Excess concentrated ammonia solution is added to a solution of nickel (II) sulfate.
129. A solution of copper (II) chloride is added to a solution of sodium sulfide.
130. A solution of tin (II) nitrate is added to a solution of silver nitrate.
131. Excess hydrobromic acid solution is added to a solution of potassium hydrogen carbonate.
132. Powdered strontium oxide is added to distilled water.
133. Carbon monoxide gas is passed over hot iron (III) oxide.